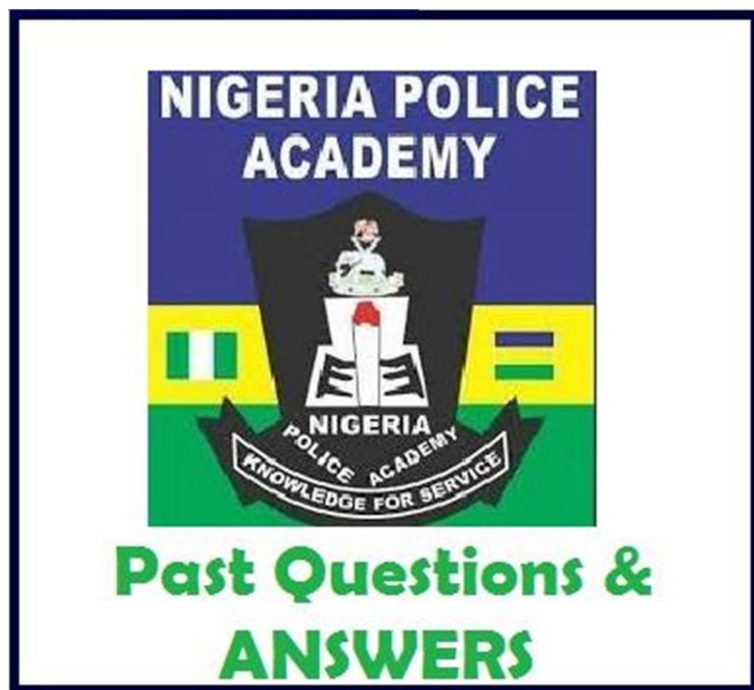


NIGERIAN POLICE ACADEMY WUDIL, KANO

Regular Course past Questions and Answers

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General Course

From the options listed A – D select the word that best fit the questions

1. How many ethnic groups do we have in Nigeria? (A) 250 ethnic groups (B) 5 ethnic groups (C) 40 ethnic groups (D) 7 ethnic groups
2. Who is the current chief of justices of Nigeria? (A) Buhari (B) President Buhari (c) Prof. Osinbajo (D) Ibrahim Tanko Muhammad
3. In Nigeria, democracy day is now celebrated on., (A) 14 April, (B) June 12 (C) 1st October (D) 27 May
4. 6. Who is the current Nigeria's Inspector General of Police is? (A) IGP Buhari (B) Adamu Mohammed (C) Lt. Cornel Buratai (D) Adamu Buhari.
5. Which African country first gained independence? (A) Nigeria on 1960 (B) U.S.A in 1874 (C) Liberia in 1847 (D) Cameron in 1960
6. Who is Nigeria's current speaker of House of Representatives? (A) Prof. Osinbajo. (B) Femi Gbajabiamila (C) Dr. Ngige (C) Hon. Bassey Albert
7. Who is Nigeria's Minister of Power? (A) President Buhari (B) Lt. Cornel Buratai (C) Sale Mamman from Taraba State (D) Adamu Mohammed
8. 14. Who was the first President of Nigeria ? (A) Chief Dr. Obasinjo (B) Dr. Nnamdi Azikiwe (C) Goodluck Jonathan (D) President Buhari
9. 15. Blue in the Nigerian Police flag represents? (A) Affection (B) Unity (C) Love, loyalty and unity (D) Peace.
10. Who was Nigeria's first inspector General of Police? (A) Obasinjo (B) Dr. U. w Jonathan (C) Lt. Buratai (D) Mr. C. W. Duncan.
11. Who is the current Sultan of Sokoto caliphate? (A) Muhammadu Stephen (B) Muhammadu Ngige (C) Muhammadu Salad Abubakar (D) Muhammadu Gregory
12. Who is the current Governor of Central Bank of Nigeria? (A) Mr. Godwin Udom (B) Godwin Emefiele (C) Dr. Jonathan (D) Dr. Peters Peters

13. Who is Nigeria's Minister of Education? (A) Adamu Adamu from Bauchi State (B) Dr. Gregory from Kaduna (C) Dr. Peters from Akwa Ibom (D) Prof. Martins for Enugu.
14. When did Nigeria break diplomatic relations with Israel? (A) 1995 (B) 1975 (C) 1972 (D) 1984
15. 29. Who is the current Senate President of Nigeria? (A) Ahmed Ibrahim Lawan (B) Abubakah Ibrahim Lawan (C) Muhammed Ibrahim Lawan (D) Ibrahim Lawan David
16. Who was the first Senate President of Nigeria? (A) Bassey Albert (B) Chief Dr. Obasinjo (C) Dr. Nnamdi Azikiwe (D) Prof. Osinbajo
17. Who was the first governor-general of colonial Nigeria? (A) Lord Lugard (B) Obama (C) Herbert Macaulay (D) Martins David
18. How many Local Government Areas are in Nigeria? (A) 189 (B) 950 (C) 752 (D) 774
19. Which state in Nigeria has the largest number of Local Government Areas? (A) Kano State (B) Akwa Ibom State (C) Delta State (D) Lagos State
20. When did the Federal Military Government abolish the four regions? (A) 1966 (B) 1960 (C) 1974 (D) 1990
21. 46. How many countries constitute the Economic Community of West African States? (A) 14 (B) 16 (C) 18 (D) 5
22. What is the ultimate aim of political parties? (A) To make Law (B) To fight Government (C) To appoint a leader (D) To acquire and exercise power.
23. How many states were created by General Yakubu Gowon in Nigeria, May 1967? (A) 12 states (B) 14 States (C) 5 States (D) 10 States.
24. Who was the first Executive President of Nigeria? (A) President Buhari (B) President Osinbajo (C) President Sanusi (D) Alhaji Shehu Shagari
25. How many regional divisions does Africa have? (A) Five Regional Divisions (B) Ten Regional Divisions (C) Six Regional Divisions (D) Seven Regional Divisions.
26. Who is the current managing director of NNPC? (A) Mr. Udom (B) Dr. Tunibu (C) Maikanti Baru (D) Engr. Martins.

27. When was the United Nations Organisation formed? (A) 25th April 1945 (B) 27 May 1960 (C) 1st October 1964 (D) 2nd January 1950.
28. The first female Brigadier in Nigeria? (A) Peter James (B) Ronke Kale (C) Benard Okorie (D) Clifford Peter
29. What is the motto of Nigeria? (A) Love, Peace and Joy (B) Unity and Faith, Peace and Progress (C) Love, Faith, Peace and Joy (D) Unity, Love, Peace and progress.
30. Who was Nigeria's First Military Head of State (A) Gen. Gowon (B) Gen. Buhari (C) Gen. Sanusi (D) Gen. Yakubu.
31. When was the Central Bank of Nigeria established? (A) 1958 (B) 1950 (C) 1995 (D) 1940.
32. When was Nigerian Airforce established? (A) 1990 (B) 1960 (C) 1964 (D) 1970.
33. When did the first world war take place? (A) 1913 – 1914 (B) 1915-2000 (C) 1910-1912 (D) 1914-1919.
34. When did the Nigerian civil war begin? (A) May 27, 1960 (B) July 6, 1967 (C) October 6, 1960 (D) December 12, 1970
35. What is the motto of the Nigerian Police? (A) Police like 50 Naira (B) Police is your friend (C) Police love you (D) Police is your enemy.
36. When did the second world war take place? (A) 1995 – 1999 (B) 1939-1945 (C) 1960 -1970 (D) 1995 – 1990.
37. When was Nigeria declared and recognized as a nation? (A) 27 May 1960 (B) 27 June 1945 (C) October 1, 1960 (D) 2nd June 1960.
38. When did the outbreak of American civil war take place? (A) 1990 – 1998 (B) 1848-1861 (C) 1914 -1915 (D) 1920 -1930
39. Who was the National president of Nigeria? (A) Goodluck Jonathan (B) Gowon (C) Obasanjo (D) Buhari
40. When did Nigeria become a British colony? (A) 1914 (B) 1920 (C) 1960 (D) 1995
41. When did the current president of Nigeria resume duties? (A) May 19, 2015 (B) June 19, 2015 (C) July 19, 2015 (A) April 19, 2015

42. What is the state of origin of the president of Nigeria? (A) Kano (B) Niger state (C) Sokoto (D) Katsina State
43. Who is the Vice President of Nigeria? (A) Oluyemi Osibanjo (B) Osibanjo (C) Oluyemi (D) Prof. Oluyemi Osibanjo.
44. Which is the highest court of law in Nigeria? (A) High Court (B) Customary Court (C) Supreme court (D) Majesty court.
45. Who designed the Nigerian flag? (A) Michael Taiwo Akinkunmi (B) Taiwo (C) Akinkunmi (D) Michael
46. What do the two horses on the Nigerian coat of arm represent? (A) Love (B) Royalty (C) Dignity (D) Peace.
47. Question: Who is the current minister of education? (A) Prof. Osinbajo (B) Prof. Bassey (C) Mr. Peters (D) Adamu Adamu
48. Question: Who is the current president of Nigeria? (A) Muhammadu Buhari (B) Prof. Yemi Osinbajo (C) Atiku (D) Peter Obi
49. After the defeat of Germany in world war 1, her former colonies were administered under the league of nation as (A) occupy territory (B) trust territories (C) crown colonies (D) protectorates
50. Where is Nigeria Police Academy Located ? (A) Kano (B) Kaduna (C) Jos (D) Sokoto

General Course Answers

1. A
2. D
3. B
4. B
5. C
6. B
7. D
8. B
9. C
10. D
11. C
12. B
13. A
14. C
15. A
16. A
17. D
18. D
19. A
20. A

21. B
22. D
23. A
24. D
25. C
26. C
27. A
28. B
29. B
30. A
31. A
32. C
33. D
34. C
35. B
36. B
37. C
38. B
39. C
40. A
41. A
42. D
43. D
44. C
45. A
46. C
47. D
48. A
49. B
50. A

ENGLISH LANGUAGE

Choose the expression or word which **BEST COMPLETES** each sentence

1. I am intent _____ continuing my course
A. on B. with C. as D. to E. at
2. _____ equations use letters to stand for numbers
A. Silmultenous B. Simultaneous
C. Simultaneous D. Simultaneous

From the options A to D, choose the expression that is **OPPOSITE IN MEANING** to the underlined word(s).

3. Elemi's sagacity contrasted sharply with his friend's _____
A. Timidity B. Wisdom C. Fluency D. Foolishness
4. Some people keep ferocious animals as pets.
A. Gentle B. Wild C. Fierce D. Domestic

From the options listed A – D select the word that best **CAPTURES THE MEANING** of the italicized part of the sentences

5. Every visitor to Calabar must visit where photographs and artifacts of early European colonial presence in Nigeria are kept
A. Archives B. Ranch C. Market D. museum

From the options A to D, choose the expression that is **NEAREST IN MEANING** to the underlined word

6. Ebire's generosity turned out to be her **Achilles' heel**
A. Strong point B. Favourite habit C. Weak point D. Less popular virtue

Choose the expression or word which **BEST COMPLETES** each sentence

7. The giant hydro-electric project is among the _____ of colonial rule in Southern Africa
A. Inheritance B. Remnants C. Legacies D. Evidence

8. Ukpabio is proficient _____ tailoring
A. With B. In C. Of D. At

Choose the word or phrase that is OPPOSITE IN MEANING to the underlined

9. The demonstration was organized by hoodlums
A. Criminals B. Activists C. Thugs D. Soldiers
10. The plaintiff convinced the court that the murder was inadvertent
A. Brutal B. Wicked C. Careless D. Premeditated
11. The bush burning festival will further renew our forest resource
A. Increase B. Reduce C. Deplete D. Remove

Choose from the options lettered A – E the one that has the correct stress with the word given. In each word only the stressed part is in CAPITAL

12. Planetarium
A. PLAN-e-tar-i-um B. plan-E-tar-i-um C. plan-e-TAR-i-um D. plan-e-tar-I-um

Choose from the options the word which has THE SAME SOUND as the underlined

13. COURTESY A. our B. Court C. Shirt D. Tour

From the options listed A – D select the word that best CAPTURES THE MEANING of the italicized part of the sentences.

14. Lecturers assisted Prof. Atahiru Jega when *Nigerians chose their political leaders*
A. election B. nomination C. appointment D. selection
15. Mother always *pays attention to details*
A. frivolous B. interesting C. rigid D. meticulous

Choose the option that BEST EXPLAINS the information conveyed in the sentences below

16. The rampage in England shows that youths react the same way under provocation
A. youths in England are very good B. youths in England are miscreants
C. youths act the same way when provoked D. youths provoke people in England

From the options choose the appropriate STRESS pattern. The stressed syllables are written in CAPITAL letters.

17. Intimacy
 A. INtimacy B. inTImacy C. intiMACy D. INTImacy

18. consideration

- A. CONsideration B. consideration C. consiDERation D. considerAtion

From the words lettered A – D choose the word that rhymes with the given word

19. purity
 A. plentiful B. security C. purify D. nonentity

From the words lettered A – D choose the word that rhymes with the given word

20. beautiful
 A. beautify B. beautification C. dutiful D. dignify

Choose from the options lettered A – E the one that has the correct stress with the word given. In each word only the stressed part is in CAPITAL

21. constitutional
 A. con-sti-tu-tion-AL B. con-sti-TU-tion-al C. CON-sti-tu-tion-al D. con-STI-tu-tion-al

22. abnormality
 A. AB-nor-ma-li-ty B. ab-nor-ma-li-TY C. ab-nor-ma-LI-ty D. ab-nor-MA-li-ty

From the options A to D, choose the expression that is NEAREST IN MEANING to the underlined word

23. The use of Latin expressions in English is now dated
 A. well-established B. historical C. old-fashioned D. popular

24. His uncle showed affected interest in his welfare.
 A. pretended B. loving C. genuine D. deep

From the options choose the expression that BEST COMPLETES each sentence

25. I am qualified for the job; ----?
 A. haven't I B. Isn't it C. aren't I D. ain't I

26. My sister along with her colleagues ----the museum today.
 A. are visiting B. have visited C. were visiting D. is visiting

Choose the word that has the SAME SOUND as the one represented by the underlined letter

27. mansion
 A. cheap B. leisure C. peace D. action

28. time
 A. print B. might C. flip D. illegal

From the options choose the expression that BEST COMPLETES each sentence.

29. Every student _____ before the principal entered the hall
 A. has arrived B. have arrived C. had arrived D. arrived

30. The workers presented _____ to the National Assembly
 A. five pages document B. five-paged document
 C. a five - paged document D. a five - page document

From the options A to D, choose the expression that is NEAREST IN MEANING to the underlined word(s).

31. A dogged student is likely to succeed
 A. Studious B. Clever C. Curious D. Determined

Choose the option that best explains the information conveyed in the sentence below

32. When I visited Okon I only gave him a congratulatory card
 A. the only person I visited was Okon and nobody else
 B. it was only Okon that I gave a congratulatory card
 C. Okon only wanted a congratulatory card
 D. All that I gave Okon when I visited him was a congratulatory card

From the options choose the expression that BEST COMPLETES each sentence

33. The police -----up a list of suspects in the recent bombings.
 A. Has drawned B. Has drawn C. have drawned D. Have drawn

Choose the word that has the same sound as the one represented by the underlined letter

34. book
 A. lamb B. flock C. slump D. club

- Choose the expression or word which best completes each sentence**

- From the options listed A – E select the word that best captures the meaning of the italicized part of the sentences

- Select the word that best captures the meaning of the italicised part of the sentence.

- Choose the expression or word which best completes each sentence.

- Fill in the blank in the following sentences making use of the best of the five options.

- After each of the following sentences, a list of possible interpretation of all or part of the sentence is given.

Choose the interpretation that you consider most appropriate for each sentence:

- Complete the following sentences with the right option from the words supplied after each sentence.

- 9

53. Astronomers keep a close watch of the night sky in order not to miss the _____ appearance of some stars
A. Parallel **B. Periodic** **C. Regular** **D. Constant**

Choose the word or phrase from options A-D which has the nearest meaning to the underlined word or words in each sentence.

54. My mother has refused to come to live in Lagos because she prefers the tranquil life in the village to the hurly burly of the city.
A. Sweet **B. Prosperous** **C. Peaceful** **D. Decent**
55. A lorry larger than an elephant was struck on the bridge
A. As large as an elephant **B. Carrying an elephant.**
C. That looked like an elephant **D. Of enormous proportions.**

Choose the option that best completes the sentence.

56. The dentist found that his patient's teeth _____
A. Have long decayed **B. Have long been decayed**
C. Have long being decayed **D. Had long decayed**
57. In order to carry out the necessary examination, the dead body was _____
A. Extracted **B. Exhumed** **C. Extradited** **D. Expelled**
58. He went abroad with a view _____ a business partner
A. To fund **B. To funding** **C. To be funding** **D. To have funded**

In question 59, which of the options express the same idea as the one in quote?

59. 'To put something aside' is to
A. Put it one's side. **B. Put it in a side pocket for future use.**
C. Keep something for some special purpose. **D. Keep in safety.**

Select the option that best explains the sentence:

60. The painting was beautifully faked
A. The painting was a good deceptive replica **B. The painting was well framed and displayed**
C. The painting was deceptively decorated **D. The painting was carefully hung**

From the alternatives provided select the one which best completes the sentence

61. If only I _____ insured? But I wasn't. But I have to pay a lot of money
A. am **B. have been** **C. had been** **D. was to be**
62. Since the petition writer did not include his name, the vice chancellor refused to act on such _____ letter
A. a spontaneous **B. an anonymous** **C. a scandalous** **D. a cowardly**
63. The first graduation ceremony of the university was attended by men from all _____ of life
A. works **B. areas** **C. walks** **D. walk**

Choose the word or phrase from options A – D which has the nearest meaning to the underlined word or words in each sentence

64. He should be able to do it alone
A. he would be able to do it alone **B. he ought to be able to do it alone**
C. he has to be able to do it alone **D. he will do it alone**
65. The old man was said to have died intestate
A. without an estate **B. in his estate**
C. without a will **D. in good state**

From the alternatives provided select the one which best completes the sentence

66. If only I _____ insured? But I wasn't. But I have to pay a lot of money
A. am **B. have been** **C. had been** **D. was to be**

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A. works **B.** areas **C.** walks **D.** walk

Choose the word or phrase from options A – D which has the nearest meaning to the underlined word or words in each sentence

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A. without an estate **B.** in his estate **C.** without a will **D.** in good state

In each of the following questions, the word in capital letter has the emphatic stress. Choose the option that best fits the expression in the sentence.

71. The man **BOUGHT** the newspaper
A. Is this the newspaper which the man bought? **B.** Did the man read the newspaper?
C. Who bought the newspaper? **D.** What did the man buy?
72. The chief **RAN** to the place
A. Where did the chief run to? **B.** Did the chief walk to the place?
C. Did the princess run to the place **D.** Who ran to the place?

From the options lettered A – D choose the word opposite in meaning to the underlined words in the following sentences

73. Eber's _____ contrasts with his brother's indolence
A. indulgence **B.** laziness **C.** labour **D.** diligence
74. The discussion ended on an amicable note even though there was grave _____ at the beginning
A. hostility **B.** incompatibility **C.** irresponsibility **D.** camaraderie

Select the word that best captures the meaning of the italicized part of the sentence

75. Some politicians in the area were said to be engaged in *double dealing*
A. eschewing **B.** effeteness **C.** duplicity **D.** dubiousn

POLAC CURRENT PAST Questions and Answers

BIOLOGY

1. The wavelike motion of the muscles of the oesophagus to push each bolus of food downwards is known as
A. Anti-peristalsis B. Digestion C. Peristalsis D. Oesophageal motion
2. Which of these diseases cannot be controlled by killing the vectors?
A. River blindness B. Malaria C. Polio D. Cholera
3. If a tall man (TT) marries a dwarf woman (tt) and they have four children, what will be the ratio of tall to dwarf children?
A. 0 tall:4 dwarf B. 3 tall:1 dwarf C. 2 dwarf:2 tall D. 4 tall:0 dwarf
4. The chromosome is found in which organelle
A. The Nucleus B. The Mitochondria C. The Vesicle D. The Cytosol
5. Photosynthetic organisms are classified as _____
A. Autotrophs B. Heterotrophs C. Herbivorous D. Omnivorous
6. Which of the organisms listed below is not a micro-organism?
A. Virus B. Tapeworm C. Coccus D. Vibrio cholera
7. The nervous system is made up of
A. Kidney, bladder and liver B. Testes and ovaries C. Brain and spinal cord D. Muscles and skeleton
8. The synthesis of protein takes place in _____
A. Golgi body B. Ribosome C. Mitochondria D. Nucleus
9. Mutations involving alterations in the genes are called
A. Gene mutations B. Gene changes C. Chromosomal mutations D. Chromosomal changes
10. Which vertebra has a projection on the centrum called odontoid process?
A. Atlas B. Axis C. Thoracic D. Lumbar
11. Deamination occurs in
A. Kidney B. Pancreas C. Spleen D. Liver
12. The ability of an organisms to live successfully in an environment is known as
A. Succession B. Resistance C. Adaptation D. Competition
13. The following are connected with the movement of a reflex action
(1) Central nervous system, (2) Muscle, (3) Skin, (4) Sensory nerve, (5) Motor nerve
Which of the following sequences indicates a correct path?
A. 1-2-3-4-5 B. 2-1-4-5-3 C. 3-4-1-2-5 D. 3-4-1-5-2
14. Which of the following statements is NOT true of symbiosis?
A. Symbionts must be living B. It is an association of 'give and take' C. The association may involve two plants D. Association between two similar species
15. The deficiency of Vitamin D leads to:
A. Scurvy B. Pellagra C. Rickets D. Beriberi
16. Hepatic portal vein is unique because it
A. Carries deoxygenated blood B. Begins and ends with capillaries C. Is the largest vein in mammals D. Carried digested food
17. The thoracic vertebrae of a mammal is characterized by the
A. Vertebrarterial canal B. Prominent neural arch C. Prominent neural process D. Prominent neural spine
18. A grasshopper respire by means of its
A. Lung-books B. Gills C. Lungs D. Tracheal tubes
19. The inactive state exhibited by an animal during hot dry seasons is termed
A. Aestivation B. Dormancy C. Resting D. Hibernation

20. Auxins are produced in the
A. petiole of leaves **B.** parenchyma of roots and shoots
C. epidermis of roots and shoots **D.** apical regions of roots and shoots
21. Which of the following methods of reproduction is common in Paramecium and Amoeba?
A. Conjugation **B.** Sexual reproduction
C. Binary Fission **D.** Budding
22. An example of cryptic coloration is the:
A. Mottled colours on moths that rest on lichens **B.** Bright colour of an insect pollinated flower
C. Green colour of a plant **D.** Bright marks on a poisonous tropical frog on variegated leaves
23. Which of the following organs is responsible for controlling the body temperature regulation and water balance in mammals?
A. Kidney **B.** Hypothalamus **C.** Parathyroid **D.** Adrenal
24. A man and his wife are both heterozygous for the sickle cell trait. The likely percentage of their offspring that will either be carriers or 'sicklers' is
A. 75% **B.** 50%
C. 25% **D.** 100%
25. The brain and the spinal cord make up the
A. Peripheral nervous system. **B.** Autonomic nervous system
C. Central nervous system. **D.** Somatic nervous
26. The hormone which regulates the amount of sugar in the blood is
A. Thyroxine **B.** Auxin
C. Insulin **D.** Adrenalin
27. Which of the following is the effect of using artificial pollination in plant breeding?
A. production of healthy crops **B.** improvement of the variety of crops
C. lengthening the maturity time **D.** making crops susceptible to diseases
28. Which of the following processes will not introduce carbon dioxide into the atmosphere?
A. breathing **B.** photosynthesis
C. Respiration **D.** Putrefaction
29. Which of the following does not contribute to the biomass in an ecosystem?
A. producer's **B.** food chain
C. consumer's **D.** micro-organisms
30. Which of these diseases cannot be prevented by immunization
A. Onchocerciasis **B.** Poliomyelitis
C. Cholera **D.** Tuberculosis
31. Which of the following specialized structures are stimulated by touch, pressure, pain, heat and cold?
A. receptors **B.** synapse **C.** cell bodies **D.** myelin
32. The process by which plants and animals are modified in structure, physiology and behavior in order to survive is known as
A. evolution **B.** adaptation **C.** succession **D.** hibernation
33. Thunderstorm can be beneficial to plants because it
A. kills pests that attack crops **B.** adds lime to the soil
C. adds nitrates to the soil **D.** makes rain water available to plants
34. When large numbers of organisms share limited space and resources, the result is
A. immigration **B.** symbiosis **C.** extinction **D.** competition
35. Even though some flowering plants contain accessory pigments which give them colors, they still contain the pigment
A. melanin **B.** chlorophyll **C.** carotene **D.** xanthophyll
36. Which structure in the maize grain stores food?
A. radicle. **B.** embryo **C.** cytoplasm **D.** endosperm
37. Which of the following statements is not associated with the theory of natural selection?
A. There is a struggle for existence **B.** There is competition among the offspring
C. The weaker offspring's are eliminated **D.** Food and other needs are abundant
38. Nitrogen fixing micro-organisms in leguminous plants live symbiotically in the
A. branch roots **B.** tap roots **C.** root nodules **D.** root hairs

39. The epidermis of the mammalian skin is an example of a tissue because the cells
A. prevent light from passing through them **B.** Have a similar structure and function
C. prevent excessive loss of water **D.** are impregnated with keratin
40. The phenomenon whereby some organisms with certain features get established in an environment is known as
A. Partial selection **B.** Artificial selection **C.** Natural selection **D.** Mutation
41. Who formulated the theory of natural selection in evolution?
A. Gregor Mendel. **B.** Jean Lamarck **C.** Mathias Scheiden **D.** Charles Darwin
42. Which of the following statements is not correct of respiration?
A. Gaseous exchange occurs by diffusion
B. Oxygen combines with hemoglobin in the respiratory surface
C. Carbon dioxide produced in the tissues is removed by the process of osmosis
D. there are no special organs of respiration in plants
43. Which of the following constitutes the main internal tissues of a leaf?
A. Cuticle **B.** Mesophyll **C.** Vascular tissue **D.** Lower epidermis
44. Which of the following is an autotrophic mode of nutrition?
A. Chemosynthesis **B.** Saprophytism
C. Parasitism **D.** Symbiosis
45. Which of the following structures differentiates an animal cell from a plant cell?
A. Ribosomes **B.** Cell membrane
C. Chloroplast **D.** Mitochondrion
46. The change in colour of the chameleon serves as a means of
A. attraction to the opposite sex **B.** repulsion of the enemy
C. a camouflage from a predator **D.** regulation of body temperature
47. At which of the following stages of mitosis do the two daughter chromosomes separate completely?
A. early prophase **B.** telophase
C. Anaphase **D.** late prophase
48. The role of dead organic matter in the soil is to
A. Make the soil black. **B.** Increase the mineral salt content.
C. Provide food for all living organisms **D.** Increase the acidity of the soil.
49. If the petals of a flowering plant are removed, which of the following processes is likely to be affected?
A. Transpiration **B.** Pollination **C.** Germination **D.** Photosynthesis
50. Which of these substances is likely to be deficient in the diet of a person having goiter?
A. Potassium **B.** Calcium **C.** Iodine **D.** Sodium
51. Banana, plantain and pineapple can be grouped together because they
A. produce small seeds **B.** are multiple fruits
C. produce suckers **D.** have runners
52. The structure in the cell that controls the movement of substances in and out of the cell is the
A. Cytoplasmic membrane **B.** Nuclear membrane
C. Cytoplasm **D.** Protoplasm
53. Which of the following statements is not true of osmotic process?
A. There must be a selectively permeable membrane
B. The two solutions must be of different concentrations initially
C. It involves only the movement of water molecules
D. The two solutions are of equal concentration at the beginning of the experiment.
54. The pulmonary artery carries
A. de-oxygenated blood from the right ventricle to the lungs
B. oxygenated blood from the right ventricle to the lungs
C. oxygenated blood from the left ventricle to the right auricle
D. de-oxygenated blood from the left ventricle to the right auricle
55. Which of the following parts of the mammalian brain is involved in taking the decision to run rather than walk?

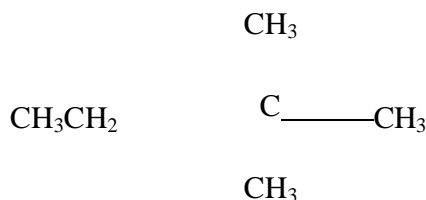
- A. Cerebellum
 C. Cranial nerves
 56. Water rises most rapidly in
 A. Sandy soil.
 C. Sandy-loam soil
 57. Euglena moves by:
 A. Whipping of its flagellum
 C. Rotating action of the flagella
 58. The following life processes are common to both plants and animal except
 A. respiration
 C. reproduction
 59. Which of the following statements is correct of hormones? Hormones are
 A. secreted into the blood through ducts
 C. inactive chemical substances in the blood stream
 60. Which of the following statements is not associated with the theory of natural selection?
 A. there is a struggle for existence
 C. the weaker offspring are eliminated
 61. Water is necessary for a germinating seed because it
 A. promotes aerobic respiration
 C. wets the soil for proper germination
 62. The farming practice by which an exhausted land is left for a number of years before cultivation is known as
 A. crop rotation
 C. mono cropping
 63. Which of the following instruments is used for determining the turbidity of the water?
 A. thermometer
 C. Rain Gauge.
 64. The surest way to combine the best qualities of both parents in the offspring is by
 A. cross-breeding
 C. selective breeding
 65. Which of the following is not true about gene mutation?
 A. it introduces new traits into a population
 C. is a source of new genes
 66. In unicellular organisms, essential nutrients can be transported directly to all parts of the body by the process of diffusion only because unicellular organisms have:
 A. A large surface area to volume ratio
 C. Permeable cell membrane
 67. Which of the following is not present in the nucleus of a cell?
 A. Chromosomes
 C. Mitochondrion
 68. The structure in the cell that controls the movement of substance in and out of the cell is the
 A. Cytoplasmic membrane
 C. Cytoplasm
 69. Which of these is not an excretory product of plant?
 A. Tannin
 C. Alkaloid
 70. Which of the following processes will not introduce carbon dioxide into the atmosphere?
 A. Breathing
 C. Respiration
 B. Medulla oblongata
 D. Cerebrum
 B. Clayey soil
 D. Loamy soil
 B. Beating of its cilia
 D. pushing out a jet of water from an organelle
 B. growth
 D. photosynthesis
 B. secreted directly into the blood stream
 D. non-specific in their mode of action
 B. there is a competition among offspring
 D. Food and other needs are abundant
 B. activates the enzymes
 D. protects the seed from desiccation.
 B. continuous cropping
 D. bush fallowing
 B. Secchi Disc.
 D. Hygrometer
 B. in breeding
 D. all of the above
 B. causes changes in the DNA
 D. always affects the chromosome number
 B. A large volume to surface area ration
 D. Their body is always immersed in the nutrient material
 B. Nucleolus
 D. Genes
 B. Nuclear membrane
 D. Protoplasm
 B. Gum
 D. Sweat
 B. Photosynthesis
 D. Putrefaction

CHEMISTRY

- _____ are allotropes of carbon
A. graphite and sulphur **B.** amorphous charcoal and coke
C. diamond and graphite **D.** carbon monoxide and coke
- A substance is said to be hygroscopic if it _____
A. does not absorb moisture **B.** is suspended in water
C. dissolves when it absorbs moistures **D.** hydrolysis in water
- $\text{CH}_3\text{CO}_2\text{H} + \text{CH}_3\text{CH}_2\text{OH}$? The functional group under and in the equation is a/an _____ group
A. Alkanol **B.** Carbonyl **C.** Keto **D.** Carboxylic acid
- The gas equation $\frac{P}{T} = \frac{P_0}{T_0} \times \frac{V}{V_0}$ is made up of Charles' Law and _____ law
A. Le Chateliers' **B.** Gay-Lussac's **C.** Boyle's **D.** Newton's
- The acid in butter is called _____ acid
A. Butanonoic **B.** Butyric **C.** Benzoic **D.** Butandioic
- The oxidation state of calcium in CaCl_2 is _____
A. +1 **B.** +2 **C.** +3 **D.** +4
- The mass number of an atom of an element is the sum of its
A. Electrons, neutrons and protons **B.** Electrons and protons
C. Protons and neutrons **D.** Valence electrons
- The existence of two or more forms of the same element in the same physical state is known as.
A. Allotropy **B.** Resonance **C.** Hybridization **D.** Isotopy
- In which of the following are radioactive isotopes used?
A. Scientific research **B.** Dating techniques
C. Treatment of cancer **D.** All of the above
- The phenomenon observed when dust particles collide randomly in a beam of sunlight is known as
A. Tyndal effect **B.** Diffusion **C.** Osmosis **D.** Brownian movement
- The main characteristic features of transition metals are that they
A. Have the same atomic size **B.** Are reducing agents
C. Forms ions easily **D.** Have variable oxidation states
- In 1898, which scientist proposed that the atom is a sphere of positively charged matter in which negatively charged electrons are embedded
A. Earnest Rutherford **B.** Max Planck **C.** J.J Thomson **D.** Robert Millikan
- Oxygen gas is collected in the laboratory by _____
A. upward displacement of air **B.** upward displacement of water
C. downward displacement of water **D.** using a gas layer
- $\text{CuO} + \text{H}_2 \rightarrow \text{Cu} + \text{H}_2\text{O}$ In the reaction above CuO is _____
A. oxidized to copper **B.** reduced to copper
C. chemically converted to copper **D.** reversed to copper
- Isotopes are atoms of the same element with _____
A. different protons **B.** mass above 50
C. inter convertible masses **D.** different masses
- A reaction is in equilibrium when _____
A. the rate of the forward reaction is equal to the rate of the reverse reaction
B. the reaction rates of the forward and backward reaction are zero
C. its rate is reversible
D. it does not produce any product again
- The boiling point of a liquid such as methanol is the temperature at which its
A. vapour pressure is equal to 1 atmosphere

- B. vapour pressure is equal to the atmospheric pressure
 C. reactants and products are at standard states
 D. vapour pressure doubles atmospheric pressure
18. Alkenes and _____ are compounds that contain a multiple bond each
 A. Alkanes B. Benzene C. Alkynes D. Benzaldehyde
19. A palm fruit dropped to the ground from the top of a tree 45m tall, how long does it take to reach the ground? (take $g=10\text{ms}^{-2}$)
 A. 1s B. 2s C. 3s D. 5s
20. _____ is due to the formation of hydrogen gas bubbles around the copper plate of a simple cell
 A. polarization B. depolarization C. local action D. amalgamation
21. Which of the following can be explained by the kinetic theory?
 I. The physical state of matter II. Diffusion of gases
 III. Melting of solid IV. Evaporation of liquids
 A. IV B. I & II C. III & IV D. I, II & III E. I, II, III & IV
22. The properties of electrovalent compounds include the following except
 A. High melting point and boiling point
 B. Conduction of electricity in the molten state
 C. High volatility at room temperature
 D. Ionization in aqueous solution
23. Which of the following statements is not corrected about electrolysis?
 A. Reduction occurs at the anode
 B. Anions migrate to the anode
 C. Positive ions migrate to the cathode
 D. Concentration affects the discharge of ions
24. Separation of mixtures of solids by physical methods can be based on differences in the following except
 A. Melting point
 B. Solubility
 C. Particle size
 D. Molar mass
25. The following are major gaseous pollutants except.
 A. CO B. CO₂ C. SO₂ D. CFC
26. Brass is an alloy containing copper and
 A. zinc B. tin C. silver D. lead
27. Which of the following reduces the activation energy of a chemical reaction?
 A. freezing mixture B. reducing agent
 C. water D. catalyst
28. In the electrolysis of brime, the anode is
 A. carbon B. platinum C. zinc D. copper
29. What current in ampere will deposit 0.27g of Aluminium in 2 hours? [Al=27, F=96500C].
 A. 3.2 B. 8 C. 0.4 D. 16
30. What is the likely formula of a compound formed between element M in group 2 and between element X in group 7?
 A. M₇X₂ B. MX₂ C. M₂X₇ D. M₂X
31. Which of the following can be explained by the kinetic theory?
 I. The physical state of matter II. Diffusion of gases III. Melting of solid IV. Evaporation of liquids
 A. IV B. I & II C. III & IV D. I, II & III E. I, II, III & IV
32. The properties of electrovalent compounds include the following except
 A. High melting point and boiling point B. Conduction of electricity in the molten state
 C. High volatility at room temperature D. Ionization in aqueous solution
 E. Decomposition of their solutions by electric current

33. Which of the following statements is not correct about electrolysis?
A. Reduction occurs at the anode **B.** Anions migrate to the anode
C. Positive ions migrate to the cathode **D.** Concentration affects the discharge of ions
E. Electrolysis conduct electric current
34. Separation of mixtures of solids by physical methods can be based on differences in the following except
A. Melting point **B.** Solubility **C.** Particle size **D.** Molar mass
35. The following are major gaseous pollutants except.
A. CO **B.** CO₂ **C.** SO₂ **D.** CFC
36. The ratio of reactants to products is 1: 3: 2 in the reaction represented by the equation below: $\text{N}_{2(g)} + 3\text{H}_{2(g)} \rightarrow 2\text{NH}_{3(g)}$ which of the following laws is demonstrated by this?
A. Boyle's law **B.** Law of multiple proportion
C. Gay Lussac's law **D.** Law of constant composition
37. PH of a 0.0001M acid is
A. 1 **B.** 2 **C.** 3 **D.** 4
- 38.38.



How many isomers of the compound above can be obtained?

- A.** 5 **B.** 4 **C.** 3 **D.** 2
39. Alkanoates are produced from alkanols by
A. Esterification **B.** Fermentation
C. Saponification **D.** Oxidation
40. Esters are employed in the following except
A. Making perfumes **B.** Nail vanisher
C. Making solvent for cellulose trioxonitrate(V) **D.** Making cement
41. What mass of anhydrous sodium trioxocarbonate(IV) is present in 500cm³ of 0.1mol dm⁻³ of the solution [Na=23, C=12, O=16]
A. 10.6g **B.** 106g
C. 5.3g **D.** 53g
42. The oxidation state of oxygen in tetraoxosulphate(IV) acid is
A. -4 **B.** +4
C. -2 **D.** +2
43. Metals of the first transition series have special properties which are different from those of groups I and II elements because they have partially filled
A. S orbital **B.** P orbital **C.** D orbital **D.** F orbital
44. Separation of different carotenes from carrot root uses a method
A. centrifugation **B.** Distillation **C.** chromatography
D. carotinzation

45.45.

Solution	W	X	Y	Z
Ph	8	12	4	2

From the table, which of the solutions W, C, Y and Z will liberate carbon (IV) oxide from a trioxocarbonate (IV) salt?

- A.** Z **B.** Y **C.** W **D.** X
46. 50cm³ of a saturated solution of KNO₃ at 40°C contained 5.05g of the salt. What is the solubility of KNO₃ at 40°C [K=39, N=14, O=16]
A. 1.0moldm⁻³ **B.** 1.5moldm⁻³ **C.** 2.0moldm⁻³ **D.** 5.0moldm⁻³

47. A brand of ink containing cobalt (III), copper (II) and iron(II) ions can best be separated into its various components by
A. fractional crystallization **B.** fractional distillation **C.** sublimation **D.** chromatography
48. If the rate law obtained for a given reaction is given as rate = $K[X]^n [Y]^m$, what is the overall order of the reaction.
A. nm **B.** $n \cdot m$ **C.** n+m **D.** n-m
49. A molecular formula shows _____ in a molecule
A. the elements present **B.** the number of atoms of each element
C. cations and anions **D.** chemical symbols and number of atoms
50. Give the total mass of copper in 1gm of copper (II) sulphate
 [Cu = 40; S = 32; O = 16]
A. 0.25g **B.** 0.50g **C.** 10g **D.** 2.5g
51. ${}_{55}^{114}\text{Cs} \longrightarrow {}^A_Z\text{E} + {}^4_2\alpha$
 Find the value of A and Z in the equation above
A. 119, 53 **B.** 110, 57 **C.** 110, 53 **D.** 110, 58
52. How many moles of H_2 molecules are needed to convert 5 mol of O_2 molecules to water?
A. 5 mol H_2 **B.** 10 mol H_2 **C.** 15 mol H_2 **D.** 20 mol H_2
53. ${}_{88}^{226}\text{Ra} \rightarrow {}_{86}^X\text{Rn} + \alpha$. What is the value of X in the nuclear reaction above?

- A. 220 B. 222 C. 226 D. 227.
54. When naphthalene on heating changes from solid state directly to the gaseous state, it undergoes
 A. sublimation B. evaporation C. combustion D. decomposition
55. Which of the following is an electrolyte?
 A. alcohol B. sodium ethanoate C. solid potassium hydroxide D. mercury
56. The equation ${}^{14}_7\text{N} + {}^4_2\text{He} \rightarrow {}^{17}_8\text{O} + {}^1_1\text{P}$ represents.
 A. nuclear fusion B. nuclear fission
 C. artificial radioactivity D. nuclear fission using positron
57. Which of the following is a general method of preparing acids?
 A. Direct combination of constituent elements B. Double decomposition involving a salt solution
 C. Reaction between an anhydride and water D. Reaction between a base and an amphoteric oxide
 E. Dissolution of hydroxides followed by neutralization
58. Monosaccharides are
 A. hydrolysable B. non-hydrolysable C. not soluble in water D. sweet but sometimes non sugary
59. The major air pollutants that can result from smoky vehicles include
 A. Acid fumes B. Hydrogen sulphide
 C. Carbon (II) oxide D. Carbon particles
60. If an element has the electronic configuration $1s^2, 2s^2, 2p^6, 3s^2, 3p^4$, it is
 A. metal B. An alkaline earth metal
 C. An S-block element D. A P-block element
61. Compounds that have the same molecular formula but different structures are said to be
 A. Allotropic B. Isotopic
 C. Polymeric D. Isomeric
62. Which of the following statements is not correct?
 A. Carbon exhibits allotropy B. Sulphur exhibits allotropy
 C. Chlorine exhibits allotropy D. Hydrogen is a gas
63. 300cm^3 of a gas has a pressure 400mmHg. If the pressure is reduced to 150mmHg. Find its volume.
 A. 700cm^3 B. 800cm^3
 C. 350cm^3 D. 112.5cm^3
64. The ideal gas equation can be written as
 A. $\frac{V_1}{T_1} = \frac{V_2}{T_2}$ B. $P_1 V_1 = P_2 V_2$
 C. $V_2 = \frac{P_1 V_1 T_1}{P_2 T_2}$ D. $P_1 V_1 T_1 = P_2 V_2 T_2$
65. Ethyne, Ethene, Cyclohexane and Propene can be classified as _____
 A. alkenes B. aromatic compounds
 C. saturated compounds D. multiple bond hydrocarbons
66. If the PH of a compound is 3.2 that compound is _____
 A. Amphoteric B. A Salt C. An Acid D. A Base
67. Which is the most ionic in the group below
 A. Cl_2 B. NaCl C. AlCl_3 D. CaCl_2
68. Hydrogenation of palm oil results in a solid compound called
 A. candle B. solid palm oil C. palm oil D. ester
69. An Amphoteric compound _____
 A. Reacts with alcohols B. Reacts both as an acid and as a base
 C. It is solid at room temperature D. has a PH of 7
70. In structural isomerism same atoms are linked to _____
 A. carbon atoms B. similar atoms
 C. different neighbouring atoms D. different functional groups

PHYSICS

1. The weakest form of bonding in materials is
A. Ionic **B.** Metallic **C.** Covalent **D.** Van der Waals
2. The period of a simple pendulum oscillating in a vacuum depends on
A. The mass of the pendulum **B.** The length of the pendulum
C. The acceleration due to gravity **D.** The volume of the pendulum
3. Which one of the following phenomenon cannot be explained by the wave theory of light?
A. Refraction **B.** Interference **C.** Diffraction **D.** Photoelectric effect
4. Which of these best describes the power of a lens
A. $P = \frac{1}{f}$ **B.** $P = \frac{1}{v}$ **C.** $P = f$ **D.** $P = v$
5. Which of the following attributes of a machine does not depend on friction?
(1) Mechanical advantage (2) Velocity ratio (3) Efficiency
A. 1 only **B.** 2 only **C.** 3 only **D.** 1 and 2 only
6. A cell is a device which converts
A. Kinetic energy to potential energy **B.** Sound energy to electrical energy
C. Chemical energy to heat energy **D.** Chemical energy to electrical energy
7. In which of the following media would sound waves travel fastest
A. kerosene **B.** alcohol **C.** water **D.** iron
8. Which of the following is not a use of plain mirrors
A. as wing mirrors on cars **B.** as a looking glass
C. in a periscope **D.** in a kaleidoscope
9. If V is the velocity of a wave, is its wavelength and T its period. The V , and T are related by the expression
A. $V = \frac{\lambda}{T}$ **B.** $V = T$ **C.** $V = \lambda T$ **D.** $V = \frac{\lambda}{T^2}$
10. Three 3.0 ohm resistors are connected in parallel. What is the equivalent resistance?
A. 9.0 ohm **B.** 1.0 ohm **C.** 6.0 ohm **D.** 3.0 ohm
11. The distance between two successive crest or trough is
A. amplitude **B.** wavelength **C.** frequency **D.** wave distance
12. An object is placed 15cm in front of a convex mirror and an image is produced 5cm behind the mirror, calculate the focal length of the mirror
A. -7.5cm **B.** 10cm **C.** 7.5cm **D.** 20cm
13. A ball is projected horizontally at 15m/s from a point 20m above a horizontal surface ($g=10\text{m/s}^2$). The magnitude of its velocity in m/s when it hits the surface is
A. 10 **B.** 15 **C.** 20 **D.** 25
14. A malaria patient has a body temperature of 39.5°C . Convert this temperature to $^\circ\text{F}$.
A. 83.2°F **B.** 103.1°F **C.** 77.4°F **D.** 147.
15. Positive charges usually move from
A. Higher to lower potential areas **B.** Lower to higher potential areas
C. North to south **D.** Higher resistance to lower resistance areas
16. A truck traveling with a velocity of 40m/s applies the brakes and comes to a halt after 20 seconds. What is distance traveled by the truck before coming to a halt
A. 40m **B.** 800m
C. 400m **D.** 10m
17. In a harmonic oscillation of a simple pendulum, one of the following statements is correct
A. The potential energy and the kinetic energy of the bob are equal at all times
B. The potential energy is equal to the kinetic energy at the central position of the oscillation
C. The potential and the kinetic energies of the bob are maximum at the maximum height of the oscillation
D. None of the above
18. If the linear expansivity of a metal rod is 4×10^{-5} per $^\circ\text{C}$, what will be the new length of the rod if it is heated from 15°C to 95°C from its original length of 20cm
A. 0.064 cm **B.** 0.64 cm
C. 20.64 cm **D.** 20.064 cm

19. The heat required to melt ice at 0°C to water at the same temperature is called
A. Apparent heat of fusion **B.** Apparent heat of vaporization
C. Latent heat of vaporization **D.** Latent heat of fusion
20. Which of the following is used in a nuclear reactor to slow down fast moving electrons?
A. Carbon dioxide gas
B. Liquid sodium metal
C. Concrete shield
D. Graphite rods.
21. Calculate the power-rating of a heating element which melted in 20g of ice at 0°C and raised the temperature of the resulting water to 60°C in 2 minutes
Specific latent heat of fusion of ice = 340 J/g
Specific heat capacity of water = $4.2\text{ J/g}^{\circ}\text{C}$
A. 5.92kw **B.** 98.6w **C.** 107w **D.** 42w
22. If a man cannot see objects very close to him the following can be used to correct the problem.
A. Convex lens **B.** Convex mirror **C.** Concave lens **D.** Concave mirror
23. when the bob of a simple pendulum is at it's highest displacement, one of the following happens:
A. kinetic energy is maximum **B.** potential energy is maximum
C. total energy is zero **D.** momentum is maximum
24. When the atmospheric pressure is very low in a given location, boiling point of water:
A. increases **B.** decreases **C.** remains **D.** cannot boil
25. Which of these statements is correct about cathode rays. They are fast moving
A. atoms **B.** neutrons **C.** electrons **D.** ions
26. Which of the following has the highest surface tension.
A. soapy water **B.** cold water
C. warm water **D.** Salt water
27. A truck traveling with a velocity of 40m/s applies the brakes and comes to a halt after 20 seconds. What is the distance traveled by the truck before coming to a halt
A. 40m **B.** 800m **C.** 400m **D.** 10m
28. If the linear expansivity of a metal rod is 4×10^{-5} per $^{\circ}\text{C}$, what will be the new length of the rod if it is heated from 15°C to 95°C from its original length of 20cm
A. 0.064 cm **B.** 0.64 cm **C.** 20.64 cm **D.** 20.064 cm
29. 44KJ of heat was used in raising the temperature of 2kg of paraffin oil from 360K to 370K. Calculate the specific heat capacity of paraffin oil
A. 2.2 J/kg/K **B.** $2.2 \times 10^3\text{ J/kg/K}$ **C.** $2.2 \times 10^5\text{ J/kg/K}$ **D.** 220 J/kg/K
30. A simple machine overcomes a load of 4000N when a force of 200N is applied. If the velocity ratio of the machine is 25, calculate the efficiency of the machine?
A. 1.25% **B.** 80% **C.** 125% **D.** 0.8%
31. A long sighted person is to read a book held at a distance of 20cm from the eyes. Which of the following will the person require to read the book with ease?
A. Nothing **B.** Concave lens **C.** Convex lens **D.** Concave mirror
32. An object at the bottom of a pool of liquid 10m depth is seen by an observer as if it is at 8m depth. What is the refractive index of the liquid
A. 0.25 **B.** 0.20 **C.** 1.25 **D.** 0.8
33. Which of the following has the highest surface tension?
A. soapy water **B.** cold water **C.** warm water **D.** Salt water
34. The emf developed in a circuit is directly proportional to the rate of change of magnetic flux. The above was a finding from
A. Maxwell **B.** Faraday **C.** Ampere **D.** Lenz
35. Which of the following is stored by a dry leclanche cell?
A. Chemical energy **B.** Solar energy **C.** Electrical energy **D.** Heat energy
36. A ball is projected horizontally from the top of a hill with a velocity of 20m/s . if it reaches the ground 4 seconds later. What is the height of the hill? ($g=10\text{m/s}^2$).
A. 200m **B.** 20m **C.** 160m **D.** 40m

37. The expansion of solids can be considered a disadvantage in
A. fire-alarm system **B.** thermostat
C. riveting of steel plates **D.** balance wheel of a watch
38. An object is placed between 2 mirrors inclined at an angle of 120° and facing each other. Determine the number of images observed in the 2 mirrors
A. 2 **B.** 3 **C.** 4 **D.** 1
39. Which of these instruments is suitable for making the most accurate measurement of the internal diameter of a test tube?
A. metre rule **B.** vernier callipers
C. micrometer screw gauge **D.** spectrometer
40. Which of the following instruments would be best suited in studying the stars?
A. periscope **B.** microscopes
C. telescopes **D.** spectroscopy
41. A gas which obeys Charles law exactly has a volume of 283cm^3 at 10°C . what is the volume at 30°C ?
A. 142cm^3 **B.** 293cm^3 **C.** 303cm^3 **D.** 566cm^3
42. Electrical resistance is a property of an electric conductor that causes electrical energy to be converted to
A. mechanical energy **B.** solar energy **C.** heat energy **D.** chemical energy
43. Which of the following has the greatest penetrating power?
A. beta particles **B.** Alpha particles **C.** gamma rays **D.** Neutrons
44. The change in direction of a wave front as a result of a change in the velocity of the wave in another medium is called
A. interference **B.** diffraction **C.** refraction **D.** reflection
45. Which of the following is a reason why concrete floor feels colder to the bare feet than a mat on the same floor during the rainy season?
A. mat is a better conductor of heat than the feet
B. mat loses heat to the bare feet at a faster rate than concrete floor.
C. mat loses heat to the bare feet while the concrete floor extracts heat from them.
D. concrete floor is a better conductor of heat than the mat.
46. A short-sighted person needs one of the following types of lens to correct the vision
A. convex lens **B.** concave lens **C.** plano-convex lens **D.** plano-concave lens
47. The motion of a body is simple harmonic if the
A. acceleration is always directed towards a fixed point **B.** path of motion is a straight line
C. acceleration is directed towards a fixed point and proportional to its distance from the point
D. acceleration is proportional to the square of the distance from a fixed point.
E. acceleration is constant and directed towards a fixed point.
48. An object is placed 25cm in front of a convex lens of focal length 10cm . The image of the object will be
A. real inverted and magnified **B.** real upright and magnified
C. real upright and diminished **D.** real inverted and diminished
49. Any successful lift-off of space-craft uses the application of
A. Newton's First Law **B.** Newton's Second Law **C.** Newton's Third Law **D.** All of the above
50. When a carpenter screws a nail into an object, the distance travelled into the object at a full circle of the head of the screw is called
A. pitch **B.** thimble **C.** wavelength **D.** moment
51. One of the following is true of Boyle's law when the pressure of a given volume of gas increases:
A. volume increases **B.** the temperature increases
C. the volume decreases **D.** the temperature decreases
52. If two resistors of 10Ω and 20Ω are connected in series and then placed in parallel with another 20Ω resistor, calculate the effective resistance of the combination
A. 50Ω **B.** 12Ω **C.** 10Ω **D.** 0.02Ω
53. Converging lenses with focal lengths 4cm , 40cm and 4m are available for use. Which of them will you use as a magnifying glass
A. 4cm **B.** 40cm **C.** 4m **D.** None of the above

54. A piece of aluminium of mass 0.5kg at a temperature of 100°C is placed in water of mass 0.4kg at a temperature of 10°C. If the final temperature of the mixture is 40°C. Find the specific heat capacity of aluminium if that of water is 4200J/Kg/°C
A. 960J/Kg/°C **B.** 908J/Kg/°C **C.** 1860J/Kg/°C **D.** 1680J/Kg/°C
55. Hardness of X-rays is a measure of its
A. Intensity **B.** Density **C.** Penetration power **D.** Picture it produces
56. The half life of a radioactive substance is 2 seconds. Calculate the decay constant
A. 0.035s **B.** 0.347s **C.** 0.151s **D.** 0.576s
57. A body moves with a constant speed but has an acceleration. This is possible if it
A. moves in a straight line **B.** moves in a circle **C.** is oscillating **D.** is in equilibrium
58. A 500kg truck moving with a velocity of 6m/s collides with a stationary 100kg car. The truck and the car move together after collision. What is their common velocity?
A. 5m/s **B.** 10m/s **C.** 30m/s **D.** 1m/s
59. The process by which a metal heated to a high temperature gives off electrons from its surface is known as:
A. radioactive emission **B.** field emission
C. photoelectric emission **D.** thermionic emission
60. A total internal reflection occurs in the glass-air boundary when the angle of incidence is
A. greater than the right angle **B.** greater than critical angle
C. greater than angle of prism **D.** greater than angle of deviation
61. At what temperature will the Celsius scale coincide with the Fahrenheit scale?
A. 4° **B.** -273° **C.** 100° **D.** -40°
62. Which of these statements is correct about cathode rays. They are fast moving
A. atoms **B.** neutrons **C.** electrons **D.** ions
63. The upthrust experienced by an object immersed in a fluid makes the object feel
A. Heavier **B.** Lighter **C.** sink lower **D.** none of the above
64. Which of the following has the highest surface tension.
A. soapy water **B.** cold water
C. warm water **D.** Salt water
65. Which of these is NOT a consequence of hydrogen bubbles covering the upper plate of a primary cell?
A. Polarization **B.** Local action
C. Generation of less current by the cell **D.** Increase in the resistance of the cell.
66. What is the current in a circuit if the charge of 240 coulombs pass each point in 2 minutes
A. 120 A **B.** 480 A **C.** 4 A **D.** 2 A
67. An electric in Nigeria has a number of 60w coloured bulb. How many can be connected to a 240V supply through a 5A fuse
A. 20 **B.** 48 **C.** 5 **D.** 4
68. A radio station broadcasts on a frequency of 100MHz. If the speed of the radio wave is 3×10^8 m/s, what is the wavelength of the radio wave from this station
A. 3×10^6 m **B.** 3×10^3 m **C.** 300m
D. 3m
69. If you need to clearly see an object that is at a very far distance, then you need to use
A. periscope **B.** microscope
C. telescope **D.** prismscope
70. Michael Faraday found that the EMF developed across a conductor is directly proportional to the
A. current applied **B.** voltage applied
C. rate of change of magnetic flux **D.** rate of change of electric flux.

ANSWERS PAGES

ENGLISH

1. A
2. D
3. D
4. A
5. D
6. C
7. C
8. D
9. B
10. D
11. C
12. C
13. C
14. A
15. D
16. C
17. A
18. D
19. B
20. C
21. B
22. D
23. C
24. A
25. C
26. D
27. D
28. B
29. C
30. D
31. D
32. D
33. D
34. D
35. B
36. C
37. A
38. B
39. D
40. D
41. B
42. D
43. C
44. C
45. D
46. D
47. A
48. A
49. C
50. A
51. C
52. B

- 53. B
- 54. C
- 55. D
- 56. D
- 57. B
- 58. B
- 59. C
- 60. A
- 61. C
- 62. B
- 63. C
- 64. A
- 65. D
- 66. C
- 67. B
- 68. C
- 69. A
- 70. D
- 71. B
- 72. B
- 73. D
- 74. B
- 75. C

BIOLOGY

- 76. C
- 1. D
- 2. D
- 3. A
- 4. A
- 5. B
- 6. C
- 7. B
- 8. A
- 9. B
- 10. D
- 11. C
- 12. D
- 13. D
- 14. C
- 15. D
- 16. D
- 17. D
- 18. A
- 19. D
- 20. C
- 21. D
- 22. B
- 23. A
- 24. C

25. C
26. B
27. B
28. B
29. A
30. A
31. B
32. C
33. D
34. B
35. D
36. D
37. C
38. B
39. C
40. D
41. C
42. B
43. A
44. C
45. C
46. C
47. B
48. B
49. C
50. C
51. A
52. D
53. A
54. D
55. A
56. A
57. D
58. B
59. D
60. B
61. D
62. B
63. A
64. D
65. A
66. C
67. A
68. D
69. B

CHEMISTRY

1. C
2. C
3. D
4. C
5. B
6. B
7. C

- 8. A
- 9. D
- 10. A
- 11. D
- 12. C
- 13. C
- 14. B
- 15. D
- 16. A
- 17. B
- 18. C
- 19. C

- 20. A
- 21. E
- 22. C
- 23. A
- 24. D
- 25. B
- 26. A
- 27. D
- 28. A
- 29. C
- 30. B
- 31. E
- 32. C

- 33. A
- 34. D
- 35. B
- 36. C
- 37. D
- 38. B
- 39. A
- 40. D
- 41. C
- 42. C
- 43. C
- 44. C
- 45. B
- 46. A
- 47. D
- 48. C
- 49. D

- 50. A
- 51. C
- 52. B
- 53. C
- 54. A
- 55. B
- 56. C
- 57. A
- 58. B
- 59. C
- 60. D
- 61. D
- 62. C
- 63. B
- 64. D
- 65. D
- 66. C
- 67. B
- 68. C
- 69. B
- 70. C

PHYSICS

- 1.
- 2. D
- 3. C
- 4. D
- 5. A
- 6. B
- 7. D
- 8. D
- 9. A
- 10. C
- 11. B
- 12. B
- 13. A
- 14. D
- 15. B
- 16. A
- 17. A
- 18. D
- 19. D
- 20. D
- 21. C
- 22. B
- 23. A
- 24. B
- 25. B
- 26. C
- 27. B
- 28. A
- 29. D

- 30. B
- 31. B
- 32. C
- 33. C
- 34. B
- 35. B
- 36. A
- 37. C
- 38. D
- 39. A
- 40. B
- 41. C
- 42. C
- 43. C
- 44. C
- 45. C
- 46. D
- 47. B
- 48. C
- 49. D
- 50. D
- 51. A
- 52. C
- 53. B
- 54. A
- 55. D
- 56. C
- 57. B
- 58. B
- 59. A
- 60. D
- 61. B
- 62. D
- 63. C
- 64. B
- 65. B
- 66. B
- 67. D
- 68. A
- 69. D
- 70. C

POLAC CURRENT PAST Questions and Answers